

*N,N'*-Bis(4-chlorobenzylidene)propane-1,2-diamineShu-Ping Yang,<sup>a\*</sup> Li-Jun Han,<sup>b</sup>  
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## Key indicators

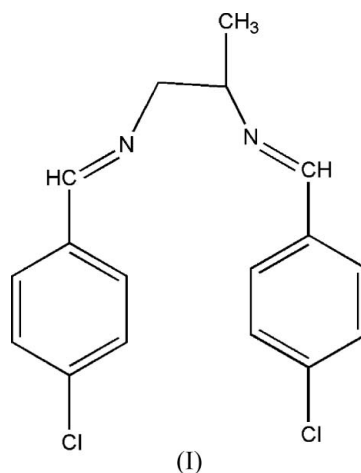
Single-crystal X-ray study  
*T* = 298 K  
Mean  $\sigma(\text{C}-\text{C}) = 0.004 \text{ \AA}$   
*R* factor = 0.045  
*wR* factor = 0.115  
Data-to-parameter ratio = 14.6For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The title compound,  $\text{C}_{17}\text{H}_{16}\text{Cl}_2\text{N}_2$ , adopts a 'vault' conformation. The molecules are linked by a pair of  $\text{C}-\text{H}\cdots\text{N}$  hydrogen bonds into a centrosymmetric dimer with an  $R_2^2(18)$  ring. These dimers are linked by a weak centrosymmetric  $\pi-\pi$  interaction into a chain along the [001] direction, and these chains are connected by van der Waals interactions, resulting in a three-dimensional network structure.

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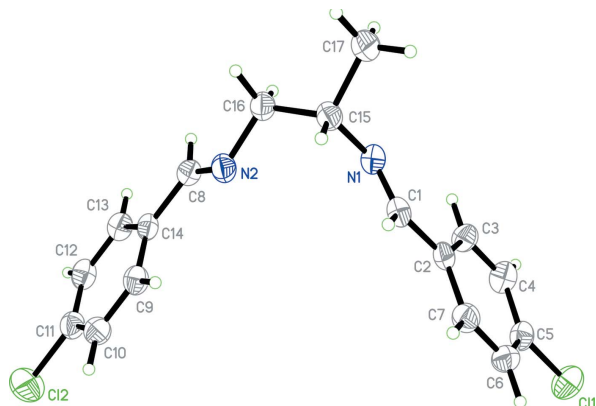
## Comment

Schiff bases are ligands of significant importance because they are potentially capable of forming stable complexes with metal ions (Johnson *et al.*, 1996; Alizadeh *et al.*, 1999), and some diamine Schiff bases have shown good anti-inflammatory activity and good analgesic activity (Sondhi *et al.*, 2006). The crystal structures of some diamine Schiff bases have been reported (Xiao & Wang, 2006; Sun *et al.*, 2004). Here, we report the crystal structure of the title diamine Schiff base, (I).

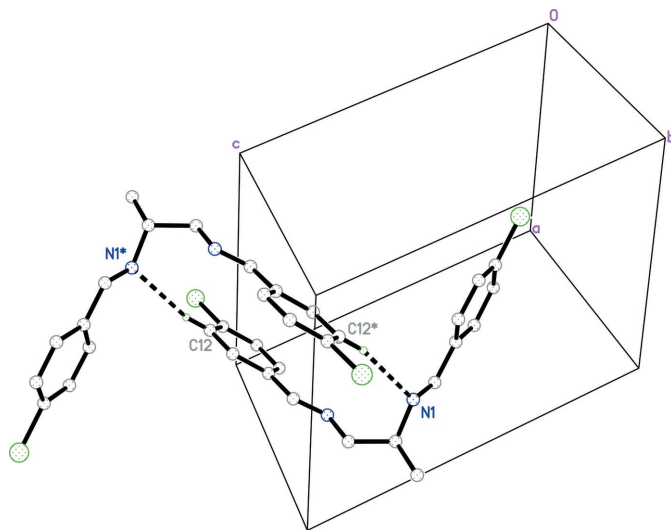


Compound (I) (Fig. 1), adopts a 'vault' conformation, with the two 4-chlorobenzyl rings located on the same side of the molecule. The two benzene rings enclose a dihedral angle of  $80.77(8)^\circ$ .

In the crystal structure of (I), the molecules are linked by a pair of  $\text{C}-\text{H}\cdots\text{N}$  hydrogen bonds into a centrosymmetric dimer with an  $R_2^2(18)$  ring (Bernstein *et al.*, 1995), centred at  $(\frac{1}{2}, \frac{1}{2}, 1)$  (Fig. 2 and Table 1). These dimers are then linked by a weak centrosymmetric  $\pi-\pi$  interaction into a chain along the [001] direction, centred at  $(\frac{1}{2}, \frac{1}{2}, \frac{1}{2})$  (Fig. 3) [ $\text{Cg}1\cdots\text{Cg}1^{\text{ii}} = 4.163 \text{ \AA}$ ;  $\text{Cg}1$  is the centroid of ring  $\text{C}2-\text{C}7$ ; symmetry code: (ii)  $1 - x, 1 - y, 1 - z$ ]. These chains are connected by van der Waals interactions, resulting in a three-dimensional network structure.



**Figure 1**  
The molecular structure of compound (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 30% probability level.



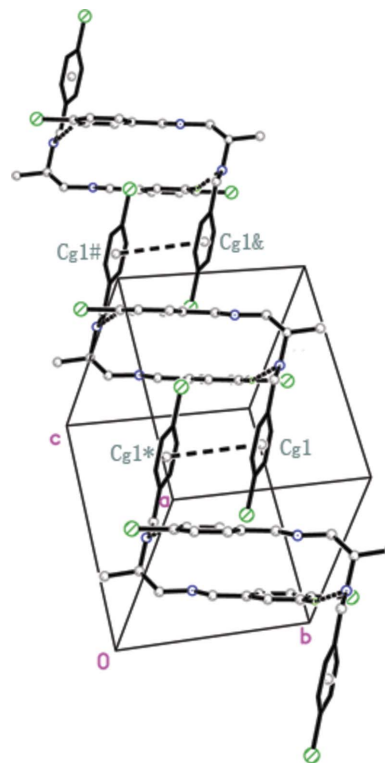
**Figure 2**  
Part of the crystal structure of (I), showing the formation of the centrosymmetric hydrogen-bonded  $R_2^2(18)$  dimer centred at  $(\frac{1}{2}, \frac{1}{2}, 1)$ . For clarity, H atoms have been omitted. [Symmetry code: (\*)  $1 - x, 1 - y, 2 - z$ ]. Dashed lines indicate hydrogen bonds.

## Experimental

A mixture of propane-1,2-diamine (1.48 g, 20 mmol) and 4-chlorobenzaldehyde (1.40 g, 10 mmol) was refluxed for about 2 h in ethanol (20 ml), and then the mixture was cooled and the products were filtered off, washed with methanol and dried. Colourless crystals of (I) suitable for X-ray structure analysis were obtained by recrystallizing the crude product from ethanol (m.p. 350–352 K).

### Crystal data

$C_{17}H_{16}Cl_2N_2$	$V = 806.8 (5) \text{ \AA}^3$
$M_r = 319.22$	$Z = 2$
Triclinic, $P\bar{1}$	$D_x = 1.314 \text{ Mg m}^{-3}$
$a = 7.731 (3) \text{ \AA}$	Mo $K\alpha$ radiation
$b = 9.301 (3) \text{ \AA}$	$\mu = 0.40 \text{ mm}^{-1}$
$c = 11.970 (3) \text{ \AA}$	$T = 298 (2) \text{ K}$
$\alpha = 81.031 (3)^\circ$	Block, colourless
$\beta = 72.354 (2)^\circ$	$0.58 \times 0.52 \times 0.19 \text{ mm}$
$\gamma = 82.339 (3)^\circ$	



**Figure 3**  
A view of part of the crystal structure of (I), showing the formation of a  $\pi$ -stacked chain of hydrogen-bonded dimers along the  $[001]$  direction. For clarity, H atoms not involved in the motif shown have been omitted. [Symmetry codes: (\*)  $1 - x, 1 - y, 1 - z$ ; (#)  $1 - x, 1 - y, 2 - z$ ; (&)  $x, y, 1 + z$ ].

### Data collection

Siemens SMART 1000 CCD area-detector diffractometer	4174 measured reflections
$\varphi$ and $\omega$ scans	2794 independent reflections
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	1704 reflections with $I > 2\sigma(I)$
$T_{\min} = 0.803, T_{\max} = 0.928$	$R_{\text{int}} = 0.020$
	$\theta_{\max} = 25.0^\circ$

### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0349P)^2 + 0.2563P]$
$R[F^2 > 2\sigma(F^2)] = 0.045$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.115$	$(\Delta/\sigma)_{\max} < 0.001$
$S = 1.06$	$\Delta\rho_{\max} = 0.22 \text{ e \AA}^{-3}$
2794 reflections	$\Delta\rho_{\min} = -0.17 \text{ e \AA}^{-3}$
191 parameters	H-atom parameters constrained

**Table 1**

Hydrogen-bond geometry ( $\text{\AA}, ^\circ$ ).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$C12-H12\cdots N1^i$	0.93	2.72	3.613 (4)	161

Symmetry code: (i)  $-x + 1, -y + 1, -z + 2$ .

All H atoms were positioned geometrically and refined as riding on their parent atoms, with  $C-H = 0.96 \text{ \AA}$  and  $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$  for methyl H atoms, and  $C-H = 0.93-0.98 \text{ \AA}$  and  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$  for all other H atoms.

Data collection: *SMART* (Siemens, 1996); cell refinement: *SAINTE* (Siemens, 1996); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997a); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997a); molecular graphics: *SHELXTL* (Sheldrick, 1997b); software used to prepare material for publication: *SHELXTL*.

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## References

- Alizadeh, N., Ershad, S., Naeimi, H., Sharghi, H. & Shamsipur, M. (1999). *Pol. J. Chem.* **73**, 915–925.
- Bernstein, J., Davis, R. E., Shimoni, L. & Chang, N.-L. (1995). *Angew. Chem. Int. Ed. Engl.* **34**, 1555–1573.
- Johnson, C. P., Atwood, J. L., Steed, J. W., Bauer, C. B. & Rogers, R. D. (1996). *Inorg. Chem.* **35**, 2602–2610.
- Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.
- Sheldrick, G. M. (1997a). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
- Sheldrick, G. M. (1997b). *SHELXTL*. Version 5.10. Bruker AXS Inc., Madison, Wisconsin, USA.
- Siemens (1996). *SMART* and *SAINTE*. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Sondhi, S. M., Singh, N., Kumar, A., Lozach, O. & Meijer, L. (2006). *Bioorg. Med. Chem.* **14**, 3758–3765.
- Sun, Y.-X., You, Z.-L. & Zhu, H.-L. (2004). *Acta Cryst.* **E60**, o1707–o1708.
- Xiao, L.-J. & Wang, D.-Q. (2006). *Acta Cryst.* **E62**, o724–o725.